metal-organic papers

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Key indicators

Single-crystal X-ray study T = 223 KMean $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.033 wR factor = 0.082 Data-to-parameter ratio = 18.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

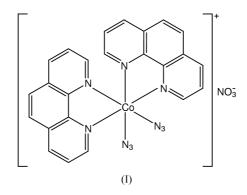
Diazidobis(1,10-phenathroline)cobalt(III) nitrate

The Co atom and nitrate anion in the title compound, $[Co(C_{12}H_8N_2)_2(N_3)_2]NO_3$, are each disposed about a twofold axis of symmetry. The Co atom exists in a distorted octahedral geometry, with the azide ligands *cis* to each other.

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Comment

The asymmetric unit in $[Co(phen)_2(N_3)_2]NO_3$, (I), comprises half each of a cation and anion, as each is located on a crystallographic twofold axis. In the cation (Fig. 1), the Co atom is hexacoordinate, existing in a distorted octahedral N6 geometry with the maximum deviation from the ideal being no more than 6.5°. The azide ligands occupy cis-positions. The Co-N distances *trans* to the azide ligands [Co-N5 = 1.9518(16) Å]are longer than the Co-N4 distances [1.9358 (12) Å] that are not (Table 1). A similar coordination geometry and pattern of Co-N bond distances was reported recently for, arguably, the most closely related structure in the literature, viz. ${Co[(NC_5H_4)NH(C_5H_4N)]_2(N_3)_2}ClO_4$ (Du *et al.*, 2001). The crystal structure features columns comprised of cations, interspersed by columns of anions, all running parallel to the c direction. Links between cations of neighbouring columns are afforded by $C12-H \cdots N3^{i}$ interactions involving the terminal N atoms of the azide ligands; $H \cdot \cdot \cdot N3^{1} = 2.55 \text{ Å}, C12 \cdot \cdot \cdot N^{1}$ 3.445 (3) Å, angle at $H = 158^{\circ}$ [symmetry code: (i) $\frac{1}{2} + x, \frac{1}{2} + y$, $z - \frac{1}{2}$]. Links between anions and cations are of the type C-H···O, the shortest of these being C9···O2ⁱⁱ is 3.142 (3) Å [symmetry code: (ii) $-\frac{1}{2} - x, \frac{1}{2} + y, z$].



Experimental

© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved Single crystals of $[Co(phen)_2(N_3)_2]NO_3$ were grown by slow diffusion, using an H-double-tube glass vessel with an aqueous solution of $Co(phen)(NO_3)_2$ (0.01 *M*) in one arm and NaN₃ (0.02 *M*) in the other. After two months, dark-brown crystals precipitated.

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Crystal data

 $\begin{bmatrix} \text{Co}(\text{C}_{12}\text{H}_8\text{N}_2)_2(\text{N}_3)_2 \end{bmatrix} \text{NO}_3 \\ M_r = 565.41 \\ \text{Orthorhombic, } Iba2 \\ a = 15.4858 (6) \text{ Å} \\ b = 9.7359 (4) \text{ Å} \\ c = 14.7769 (8) \text{ Å} \\ V = 2227.89 (17) \text{ Å}^3 \\ Z = 4 \\ D_x = 1.686 \text{ Mg m}^{-3} \end{bmatrix}$

Data collection

Bruker AXS SMART CCD diffractometer ω scans Absorption correction: multi-scan (*SADABS*, Bruker, 2000; Blessing, 1995) $T_{min} = 0.739, T_{max} = 0.960$ 8877 measured reflections

Refinement

Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.033$
$wR(F^2) = 0.082$
S = 1.00
3200 reflections
178 parameters
H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

Co-N1	1.9320 (18)	O2-N6	1.240 (3)
Co-N4	1.9358 (12)	N1-N2	1.198 (2)
Co-N5	1.9518 (16)	N2-N3	1.152 (2)
O1-N6	1.233 (6)		
N1-Co-N4	90.16 (7)	N4-Co-N4 ⁱ	179.79 (12)
N1-Co-N5	173.57 (8)	N4-Co-N5 ⁱ	96.06 (7)
N1-Co-N1 ⁱ	93.80 (12)	N5-Co-N5 ⁱ	88.68 (9)
N1-Co-N4 ⁱ	89.69 (7)	Co-N1-N2	120.19 (15)
N1-Co-N5 ⁱ	89.06 (6)	N1-N2-N3	175.6 (2)
N4-Co-N5	84.09 (7)		

Mo $K\alpha$ radiation

reflections

 $\theta = 2.5 - 29.0^{\circ}$ $\mu = 0.83 \text{ mm}^{-1}$

T = 223 (2) K

Block, brown $0.23 \times 0.23 \times 0.05 \text{ mm}$

 $R_{\rm int} = 0.037$

 $\theta_{\rm max} = 30.0^{\circ}$

 $h = -16 \rightarrow 21$

 $k=-13\rightarrow 11$

 $l=-20\rightarrow 20$

 $(\Delta/\sigma)_{\rm max} < 0.001$

 $\Delta \rho_{\rm max} = 0.56 \ {\rm e} \ {\rm \AA}^2$

 $\Delta \rho_{\rm min} = -0.25 \text{ e} \text{ Å}^{-3}$

1511 Friedel pairs Flack parameter = -0.033 (14)

Cell parameters from 3743

3200 independent reflections

 $w = 1/[\sigma^2(F_o^2) + (0.0514P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

-3

Absolute structure: Flack (1983),

2868 reflections with $I > 2\sigma(I)$

Symmetry code: (i) -x, -y, z.

The H atoms were included in the riding-model approximation, with C-H distances of 0.94 Å and $U_{iso}(H) = 1.2U_{ea}(\text{parent atom})$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SHELXTL* (Bruker, 2000); program(s) used to solve structure: *PATTY* in *DIRDIF*92 (Beurskens *et al.*, 1992); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL*97.

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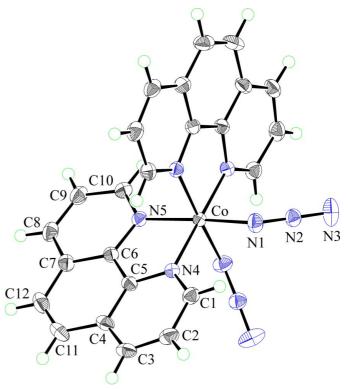


Figure 1

The molecular structure and crystallographic numbering scheme for the cation in $[Co(phen)_2(N_3)_2]NO_3$. The unlabelled half of the molecule is generated by the symmetry operation (-x, -y, z). Displacement ellipsoids are drawn at the 50% probability level (Johnson, 1976).

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